

## Progressive Education Society's Modern College of Arts, Science & Commerce Ganeshkhind, Pune – 16 (Autonomous)

End Semester Examination: MAR / APR 2025 Faculty: Science and Technology

Program: (BScGen03) Semester: VI SET: A
Program (Specific): General B.Sc. Course Type: DSEC

Class: T.Y. B.Sc.

Max. Marks: 35

Name of the Course: Organic Chemistry-II

Course Code:24-CH-607 Time: 2Hr

Paper:

#### **Instructions to the candidate:**

- 1) There are 4 sections in the question paper. Write each section on separate page.
- 2) All Sections are compulsory.
- *3)* Figures to the right indicate full marks.
- 4) Draw a well labelled diagram wherever necessary.

#### Q1) Define or Explain the following. (Any 5)

5

- a) Define chromophore.
- b) Define chemical shift.
- c) Explain the term Auxochrome with suitable example.
- d) Explain  $\pi$ - $\pi$ \* transitions with suitable example.
- e) Define coupling constant?
- f) Explain the coupling constant.

#### Q2) Very short answer questions (Attempt any 4/6)

4

- a) What is spectroscopy?
- b) Calculate the fundamental modes of vibrations in CO<sub>2</sub>.
- c) Express  $2\mu$  in cm<sup>-1</sup>.
- d) Calculate the fundamental modes of vibrations in NH<sub>3</sub>
- e) How many sets of protons are present in  $CH_3 CH_2 O CH_3$
- **f)** Draw chair conformation of cyclohexane.



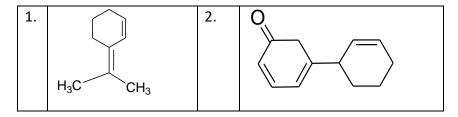
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#### Q3) Short answer questions (Attempt any 4/6)

8

a) Calculate the  $\lambda$ max values of the following compounds



### b) How will you distinguish the following pairs by I.R. spectroscopy?

- c) How IR spectroscopy is useful for the determination of hydrogen bonding in a molecule?
- d) Which are the sub regions from the view point of interpretation of IR-Spectrum?
- e) Write short note on types of vibrations.
- f) Give applications of UV spectroscopy.

#### Q4) Short answer questions (Attempt any 4/6)

8

- a) Write note on shielding and deshielding.
- b) Aniline shows blue shift in acidic medium. Explain.
- c) How many types of protons are present in the following compounds?

$$\begin{array}{|c|c|c|c|c|c|}\hline 1. & OH & 2. & CH_3 & CH_3 \\ \hline H_3C & CH_3 & H_2 & H_2 & H_2 & CH_3 \\ \hline \end{array}$$

- d) How many sets of protons are present in ethanol.
- e) What is hypsochromic shift?
- f) How IR spectroscopy is useful for the determination of hydrogen bonding in a molecule?



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## Q5) Attempt any two of the following (2/4)

10

- a) What is bathochromic shift? P-nitrophenol Shows red shift in alkaline medium explain.
- **b)** TMS is used as internal standard in NMR spectroscopy. Why?
- **c)** Propose structures for the compounds from the following spectroscopic data. Justify your answer of following compound.

M F C<sub>4</sub>H<sub>8</sub>O

IR.: 1715 cm<sup>-1</sup>

PMR: (a) Triplet at 1.07  $\delta(3H)$ 

(b) Singlet at 2.12  $\delta(3H)$ 

(c) Quartet at 2.48  $\delta(2H)$ 

d) Explain factors affecting on IR frequencies.

Page: 3/3